

Wave mechanics

Planck: $E = h\nu = \hbar\omega$

($\omega = 2\pi\nu$)
"angular frequency"

de Broglie: $p = \frac{h}{\lambda} = \hbar k$

($k = \frac{2\pi}{\lambda}$)
"wave number"

In terms of their complementary wave/particle properties, the photon and material particles, such as the electron, are

- quite similar. Thus, we (2) will endeavor to write down a wave equation for electrons, similar to the E→M wave equation:

$$\left(\nabla^2 - \frac{1}{c^2} \frac{\partial^2}{\partial t^2} \right) \vec{A}(\vec{r}, t) = 0,$$

- in free space. Here \vec{A} is the vector potential.

A plane-wave solution has the form:

$$\vec{A}(\vec{r}, t) = \vec{E} e^{i(\vec{k} \cdot \vec{r} - \omega t)}$$

- plugging this into the wave

• equation gives

$$\left(-\vec{k}^2 + \frac{\omega^2}{c^2} \right) \vec{E} = 0$$

or $\omega = c|\vec{k}|$. For a free particle of mass m , one has (neglecting relativistic effects)

$$E = \frac{\vec{p}^2}{2m}$$

$$\hbar\omega = \frac{\hbar^2 \vec{k}^2}{2m}$$

A plane wave

$$\psi(\vec{r}, t) = A e^{i(\vec{k} \cdot \vec{r} - \omega t)}$$

• for such a particle must

- Satisfy a different type of wave equation. Notice that

$$\nabla \psi = i \vec{k} \psi$$

$$\frac{\partial \psi}{\partial t} = -i \omega \psi$$

- $\vec{p} \psi = \hbar \vec{k} \psi = \frac{\hbar}{i} \nabla \psi$

$$E \psi = \hbar \omega \psi = i \hbar \frac{\partial \psi}{\partial t}$$

$$\left(E - \frac{p^2}{2m} \right) \psi = 0$$

- $\left[i \hbar \frac{\partial}{\partial t} - \left(\frac{\hbar}{i} \nabla \right)^2 / 2m \right] \psi = 0$

$$i \hbar \frac{\partial \psi}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \psi$$

- In general, for a particle $\lfloor 5$ in an external potential $V(\vec{r}, t)$, one has

$$E = \frac{\vec{p}^2}{2m} + V(\vec{r}, t).$$

- We postulate that the quantum mechanical wave function $\psi(\vec{r}, t)$ of a particle obeys the

Schrödinger equation (1)

$$i\hbar \frac{\partial \psi(\vec{r}, t)}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \psi(\vec{r}, t) + V(\vec{r}, t) \psi(\vec{r}, t).$$

- The energy carried by \vec{A} an E-M wave is proportional to the time-average of the square of the field, and hence to $|\vec{A}|^2$.

- Since $E = nh\nu$, this implies that the # of photons is proportional to $|\vec{A}|^2$. Indeed, the probability to find a photon at a given place and time is proportional to

- $$P(\vec{r}, t) \propto |\vec{A}(\vec{r}, t)|^2$$

• By analogy, we assert, [7
following Max Born, that
the probability to observe
a material particle is

$$P(\vec{r}, t) \propto |\Psi(\vec{r}, t)|^2$$

• The total probability should
be unity:

$$1 = \int d^3r |\Psi(\vec{r}, t)|^2$$

• If $\Psi(\vec{r}, t)$ satisfies this
equation, it is said to
be normalized.

- Since the Schrödinger equation is linear in ψ , we have the important
- (8)

Superposition principle:

If ψ_1 satisfies Eq. (1)

and ψ_2 satisfies Eq. (1),

then so does

$$\phi(\vec{r}, t) = a\psi_1(\vec{r}, t) + b\psi_2(\vec{r}, t)$$

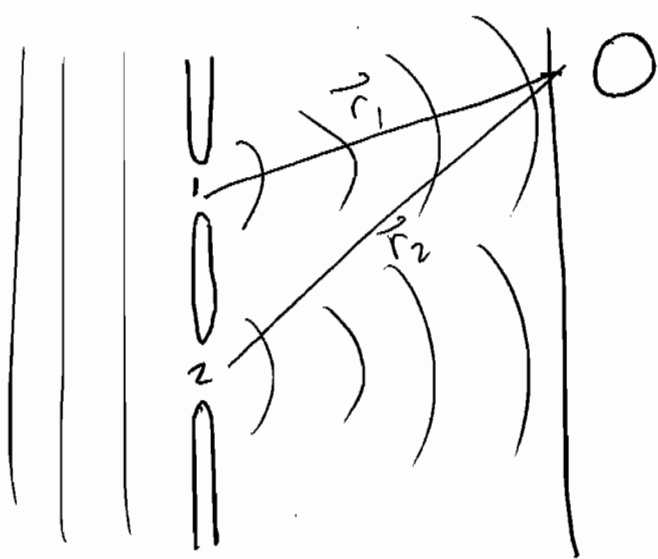
Normalization requires

$$1 = \int d^3r |\phi(\vec{r}, t)|^2$$

$$= |a|^2 \int d^3r |\psi_1(\vec{r}, t)|^2 + |b|^2 \int d^3r |\psi_2(\vec{r}, t)|^2 + a^*b \int d^3r \psi_1^* \psi_2 + a b^* \int d^3r \psi_1 \psi_2^*$$

- If $\int d^3r \psi_1^* \psi_2 = 0$, [9]
 then ψ_1 & ψ_2 are said to be "orthogonal."
 In that case, we must have $|a|^2 + |b|^2 = 1$.

Example 2-slit experiment



$$P_1 = |\psi_1|^2$$

$$P_2 = |\psi_2|^2$$

$$P_{12} = |\psi_1 + \psi_2|^2$$

$$\begin{aligned}
 P_{12} &= (\psi_1^* + \psi_2^*)(\psi_1 + \psi_2) \\
 &= \psi_1^* \psi_1 + \psi_2^* \psi_2 + \psi_1^* \psi_2 \\
 &\quad + \psi_2^* \psi_1
 \end{aligned}$$

$$\bullet P_{12} = P_1 + P_2 + \Delta P$$

(10)

$$\Delta P = \psi_1^* \psi_2 + \psi_1 \psi_2^*$$

Suppose $\psi_1 = A e^{i\vec{k} \cdot \vec{r}_1} = \sqrt{P_1} e^{i\vec{k} \cdot \vec{r}_1}$,

$$\psi_2 = B e^{i\vec{k} \cdot \vec{r}_2} = \sqrt{P_2} e^{i\vec{k} \cdot \vec{r}_2}$$

Then
$$\Delta P = \sqrt{P_1 P_2} \left(e^{-i\vec{k} \cdot (\vec{r}_1 - \vec{r}_2)} + e^{i\vec{k} \cdot (\vec{r}_1 - \vec{r}_2)} \right)$$

$$= \sqrt{P_1 P_2} 2 \cos[\vec{k} \cdot (\vec{r}_1 - \vec{r}_2)]$$