

Wave mechanics

Planck:  $E = h\nu = \hbar\omega$

$(\omega = 2\pi\nu)$   
"angular frequency"

de Broglie:  $p = \frac{h}{\lambda} = \hbar k$

$(k = \frac{2\pi}{\lambda})$

"wave number"

In terms of their complementary wave/particle properties, the photon and material particles, such as the electron, are

quite similar. Thus, we (2  
will endeavor to write down  
a wave equation for electrons,  
similar to the E→M wave  
equation:

$$\left( \nabla^2 - \frac{1}{c^2} \frac{\partial^2}{\partial t^2} \right) \vec{A}(\vec{r}, t) = 0,$$

in free space. Here

$\vec{A}$  is the vector potential.

A plane-wave solution has  
the form:

$$\vec{A}(\vec{r}, t) = \vec{E} e^{i(\vec{k} \cdot \vec{r} - \omega t)}$$

plugging this into the wave

equation gives

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$$\left( -\vec{k}^2 + \frac{\omega^2}{c^2} \right) \vec{E} = 0$$

or  $\omega = c|\vec{k}|$ . For a

free particle of mass  $m$ ,

one has (neglecting relativistic effects)

$$E = \frac{\vec{p}^2}{2m}$$

$$\hbar\omega = \frac{\hbar^2 \vec{k}^2}{2m}$$

A plane wave

$$\psi(\vec{r}, t) = A e^{i(\vec{k} \cdot \vec{r} - \omega t)}$$

for such a particle must

Satisfy a different type 4  
of wave equation. Notice  
that

$$\nabla \psi = i \vec{k} \psi$$

$$\frac{\partial \psi}{\partial t} = -i \omega \psi$$

$$\vec{p} \psi = \hbar \vec{k} \psi = \frac{\hbar}{i} \nabla \psi$$

$$E \psi = \hbar \omega \psi = i \hbar \frac{\partial \psi}{\partial t}$$

$$\left( E - \frac{p^2}{2m} \right) \psi = 0$$

$$\left[ i \hbar \frac{\partial}{\partial t} - \left( \frac{\hbar}{i} \nabla \right)^2 / 2m \right] \psi = 0$$

$$i \hbar \frac{\partial \psi}{\partial t} = - \frac{\hbar^2}{2m} \nabla^2 \psi$$

In general, for a particle [5]  
in an external potential  
 $V(\vec{r}, t)$ , one has

$$E = \frac{\vec{p}^2}{2m} + V(\vec{r}, t).$$

We postulate that the  
quantum mechanical wave  
function  $\psi(\vec{r}, t)$  of a  
particle obeys the

Schrödinger equation (1)

$$i\hbar \frac{\partial \psi(\vec{r}, t)}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \psi(\vec{r}, t) + V(\vec{r}, t) \psi(\vec{r}, t).$$

The energy carried by  $\vec{A}$  an E & M wave is proportional to the time-average of the square of the field, and hence to  $|\vec{A}|^2$ .

Since  $E = nh\nu$ , this implies that the # of photons is proportional to  $|\vec{A}|^2$ . Indeed, the probability to find a photon at a given place and time is proportional to

$$P(\vec{r}, t) \propto |\vec{A}(\vec{r}, t)|^2$$

By analogy, we assert, [7]  
following Max Born, that  
the probability to observe  
a material particle is

$$P(\vec{r}, t) \propto |\Psi(\vec{r}, t)|^2 \quad (\text{define } |\Psi|^2)$$

The total probability should  
be unity:

$$1 = \int d^3r |\Psi(\vec{r}, t)|^2$$

If  $\Psi(\vec{r}, t)$  satisfies this  
equation, it is said to  
be normalized.

Since the Schrödinger equation is linear in  $\psi$ , we have the important

Superposition principle:

If  $\psi_1$  satisfies Eq. (1) and  $\psi_2$  satisfies Eq. (1), then so does

$$\phi(\vec{r}, t) = a\psi_1(\vec{r}, t) + b\psi_2(\vec{r}, t)$$

Normalization requires

$$\begin{aligned} 1 &= \int d^3r |\phi(\vec{r}, t)|^2 \\ &= |a|^2 \int d^3r |\psi_1(\vec{r}, t)|^2 + |b|^2 \int d^3r |\psi_2(\vec{r}, t)|^2 \\ &\quad + a^*b \int d^3r \psi_1^* \psi_2 + a b^* \int d^3r \psi_1 \psi_2^* \end{aligned}$$

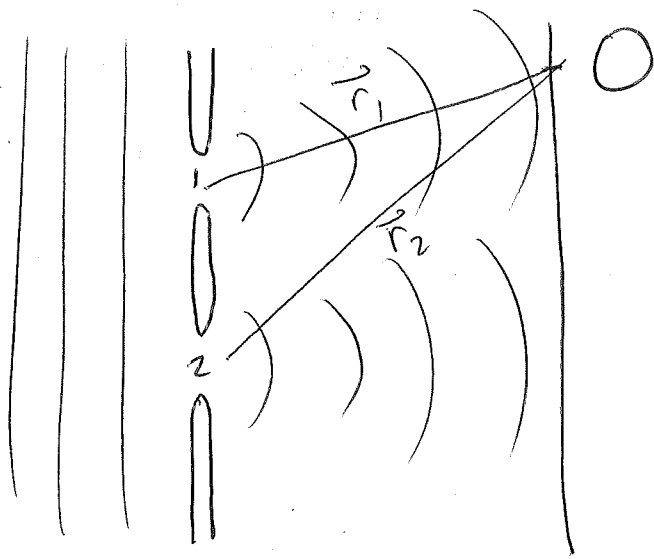


If  $\int d^3r \psi_1^* \psi_2 = 0$ , 9

then  $\psi_1$  &  $\psi_2$  are said to be "orthogonal."

In that case, we must have  $|a|^2 + |b|^2 = 1$ .

Example 2-slit experiment



$$P_1 = |\psi_1|^2$$

$$P_2 = |\psi_2|^2$$

$$P_{12} = |\psi_1 + \psi_2|^2$$

$$P_{12} = (\psi_1^* + \psi_2^*)(\psi_1 + \psi_2)$$

$$= \psi_1^* \psi_1 + \psi_2^* \psi_2 + \psi_1^* \psi_2 + \psi_2^* \psi_1$$

(10)

$$P_{12} = P_1 + P_2 + \Delta P$$

$$\Delta P = \psi_1^* \psi_2 + \psi_1 \psi_2^*$$

Suppose  $\psi_1 = A e^{i\vec{k} \cdot \vec{r}_1} = \sqrt{P_1} e^{i\vec{k} \cdot \vec{r}_1}$

$$\psi_2 = B e^{i\vec{k} \cdot \vec{r}_2} = \sqrt{P_2} e^{i\vec{k} \cdot \vec{r}_2}$$

Then 
$$\Delta P = \sqrt{P_1 P_2} \left( e^{-i\vec{k} \cdot (\vec{r}_1 - \vec{r}_2)} + e^{i\vec{k} \cdot (\vec{r}_1 - \vec{r}_2)} \right)$$

$$= \sqrt{P_1 P_2} 2 \cos[\vec{k} \cdot (\vec{r}_1 - \vec{r}_2)]$$